

**ME 227.3 Thermodynamics I**  
**Department of Mechanical Engineering**  
**Midterm Examination**  
**October 26, 2005**

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- Time: 1.5 hours
- Total marks: 50
- Calculators allowed.
- Closed book exam.
- Formula sheet supplied.
- This exam contains THREE questions.
- No Walkmans, Discmans, IPODs or similar devices allowed.
- No PDAs or laptops allowed.
- No cell phones allowed.

(10) 1. Define the following terms.

- (a) Adiabatic
- (b) Compressed Liquid
- (c) Critical Point
- (d) Cycle
- (e) Quality

(20) 2. A piston-cylinder device contains air ( $R = 286.9 \text{ J}/(\text{kg} \cdot \text{K})$ ) which undergoes three processes to form a cycle. The first process is a constant pressure expansion, the second process occurs at constant volume, and the third process is isothermal at 350 K. The maximum volume during the cycle is 5 times the minimum volume. Calculate the net work (in kJ/kg) for the cycle. Calculate the heat transfer for each process (also in kJ/kg).

(20) 3. Steam enters a turbine at 100 bar, 520°C and a mass flow rate of 5 kg/s. It leaves the turbine at 1 bar and enters a condenser where heat is removed at a rate of 10 MW. It leaves the condenser as a saturated liquid. Determine the power produced by the turbine and the quality at the turbine exit.